

## Erratum

U. Stuhr *et al.*, Hydrogen diffusion in f.c.c.  $\text{TiH}_x$ , *J. Less-Common Met.*, 172–174 (1991) 678–684.

page 679, eqn. 2 should appear as:

$$\Gamma(\mathbf{Q}, \omega) = \frac{6D(c \rightarrow 0)}{a^2} (1-c)f(c) \left( 1 - \frac{1}{3} \sum_{i=1}^3 \cos(\mathbf{Q} \cdot \mathbf{a}_i) \right)$$

page 680, line 14:

the ratio should appear as:

$$\Gamma(\mathbf{Q}/D(c))$$

page 681, the relation should appear as:

$$D(c) = D_0 \exp(-E/k_B T)$$

page 682, Figures 3 and 4 should appear as follows (the correct figure captions appear below the figures):

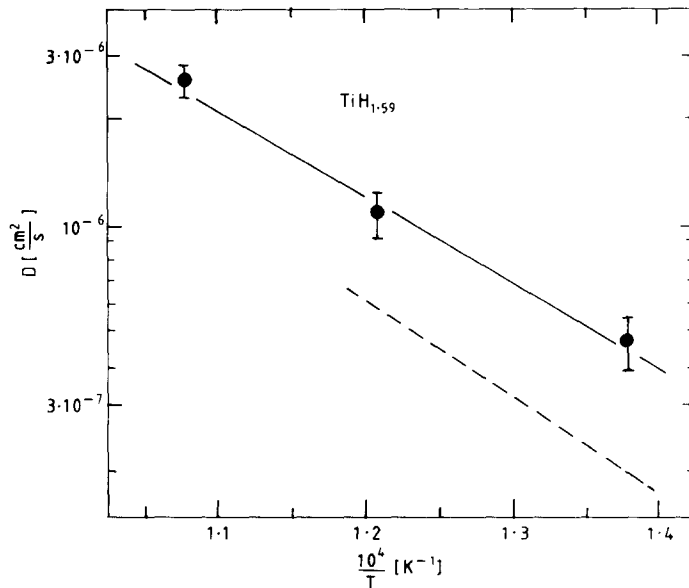


Fig. 3. Self-diffusion coefficient  $D(c)$  measured from the  $\text{TiH}_{1.59}$  sample in a semilogarithmic plot vs. reciprocal temperature  $T$ . The solid line shows an Arrhenius relation  $D(c) = D_0 \exp(-E/k_B T)$  with  $E = (0.49 \pm 0.08)$  eV and  $D_0 = (1.3^{+2.0}_{-0.3}) 10^{-3} \text{ cm}^2 \text{ s}^{-1}$ . The broken line describes NMR (pulsed field gradient) results for  $D(c)$  of a  $\text{TiH}_{1.55}$  sample [13, 14].

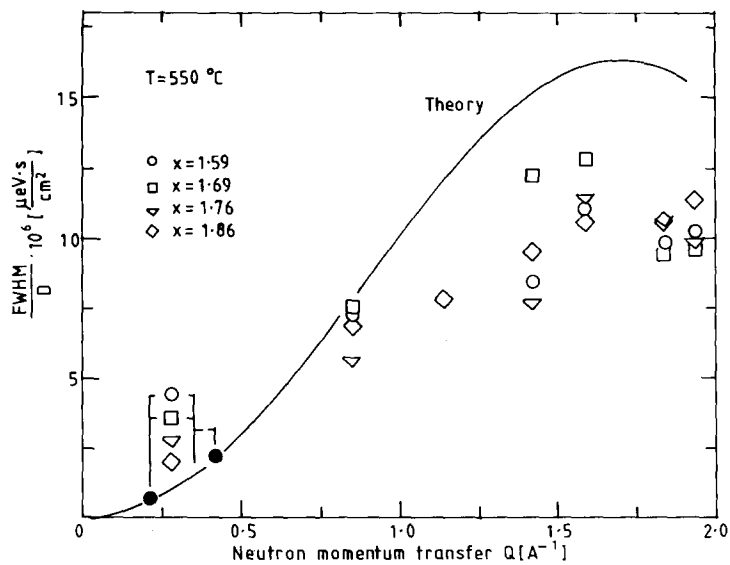


Fig. 4. The ratio  $2\hbar\Gamma(Q)/D(c)$  between linewidth  $2\hbar\Gamma(Q)$  and self-diffusion coefficient  $D(c)$  in a plot *vs.*  $Q$ . The data points result from measurements carried out at 550 °C on four TiH<sub>x</sub> samples ( $1.59 \leq x \leq 1.86$ ). The experimental accuracy of the data points ranges from 20 to 35% (except for the two lowest  $Q$  values where it is extremely small; see the text). The quoted accuracies follow as the sum of the errors for  $D(c)$  (10–20%) and  $\Gamma(Q)$  (10–15%). The solid line shows the ratio  $2\hbar\Gamma(Q)/D(c)$  as theoretically given by (5).